

MARK SCHEME for the May/June 2007 question paper

0620 CHEMISTRY

0620/06

Paper 6 (Alternative to Practical), maximum raw mark 60

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- 1 (a) A thermometer (1)
 B beaker (1)
 C tripod (1) [3]
- (b) to cool / condense the vapour (1) [1]
- (c) measure the boiling point (1) [1]
- [Total: 5]
- 2 (a) Correct indication of electrodes (1) [1]
- (b) bubbles / fizz / effervescence (1) / green gas / level of liquid falls (1)
 bulb lights up (1) max 2 [2]
- (c) (i) chlorine / Cl_2 (1) [1]
 (ii) litmus paper / indicator (1) bleaches (1) [2]
- [Total: 6]
- 3 (a) does not dissolve in solvent / interfere with results owtte (1) [1]
- (b) 1 and 3 (1) [1]
- (c) sample 4 (1)
 two spots present (1) [2]
- (d) to show position of the acids / spots (1) [1]
- [Total: 5]

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4 Table of results

Experiment 1

final reading box correctly completed, 39.2 (1)

Experiment 2

final reading box correctly completed (1)

differences completed correctly, 39.2 (1) and 20.6 (1) [4]

(a) as an indicator owtte [1]

(b) (i) Experiment 1 (1) [1]

(ii) more in Experiment 1 / greater volume (1) [1]

(iii) solution **A** more concentrated / stronger than **B** (1) approx $\times 2$ (1) [2]

(c) 10.3 (1) cm^3 / ml / cc (1) [2]

(d) change e.g. repeat titrations (1)
 explanation e.g. average reading more accurate (1) [2]

[Total: 13]

5 (c) bubbles / fizz (1) limewater (1) milky (1) [3]

(d) yellow (1) precipitate (1) [2]

(f) carbon dioxide (1) [1]

(g) ammonia (1) [1]

(h) iron (1) (II) (1) ammonium (1) sulphate (1) [4]

[Total: 11]

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6 table correctly completed

catalyst W	catalyst X
0	0
16	29
32	34
36	36
37	37
37	37
all correct (3)	-1 each incorrect

[3]

- (b) graph
 choice of suitable scale for y-axis (1)
 all points correctly plotted (3)
 smooth curves (1) labelled (1)

[6]

- (c) solid **X** (1)
 faster reaction / more gas given off at 20/40 s (1)

[2]

- (d) same volume of hydrogen peroxide used in both experiments (1)

[1]

- (e) line sketched on grid with steeper slope than for catalyst **X** at 25°C (1)
 levelling out at same level (1)

[2]

[Total: 14]

- 7 (a) initial temperature of cold water or cement (1)
 add cement (1)
 using thermometer / in beaker etc. (1)
 measure temperature (1)
 temperature rise (2) max 4

[4]

NB

no water = 0
 no cement = 0
 use of heat = 0
 wrong chemicals = 0
 would not work = 0

- (b) sodium hydroxide (1) white precipitate (1)
 or flame test (1) red (1)

[2]

[Total: 6]